

# Bond switching from two- to three-dimensional polymers of C<sub>60</sub> at high pressure

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**Abstract:** In situ high pressure x-ray diffraction experiments revealed that a transformation from the two-dimensional (2D) tetragonal C<sub>60</sub> polymer to a three-dimensional (3D) polymer takes place via a highly anisotropic deformation of C<sub>60</sub> molecules along the c axis, as an irreversible first-order transformation above 20 GPa. In the 3D polymer phase, the 2+2 bonds remain in the 2D plane, while neighboring layers are connected by the 3+3 bonds. The bulk modulus of the 3D polymer was 407 GPa, being slightly smaller than that of diamond.

**Index Keywords:** alcohol; carbon; diamond; helium; methanol; polymer; anisotropy; article; chemical bond; chemical modification; chemical structure; molecule; polymerization; pressure; X ray diffraction

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References:

1. Makarova, T.L., (2001) *Nature* (London), 413, p. 716
2. Sundqvist, B., (1999) *Adv. Phys.*, 48, p. 1
3. Iwasa, Y., (1994) *Science*, 264, p. 1570
4. Nunez-Regueiro, M., (1995) *Phys. Rev. Lett.*, 74, p. 278
5. Blank, V.D., (1994) *Phys. Lett. A*, 188, p. 281
6. Brashkin, V.V., Lyapin, A.G., Popova, S.V., (1996) *JETP Lett.*, 64, p. 802
7. Okada, S., Saito, S., Oshiyama, A., (1999) *Phys. Rev. Lett.*, 83, p. 1986
8. Meletov, K.P., (2001) *Chem. Phys. Lett.*, 341, p. 435
9. Chen, X., (2002) *Chem. Phys. Lett.*, 356, p. 291
10. Chen, X., Yamanaka, S., (2002) *Chem. Phys. Lett.*, 360, p. 501
11. Narymbetov, B., (2003) *Chem. Phys. Lett.*, 367, p. 157
12. Murnaghan, F.D., (1947) *Proc. Natl. Acad. Sci. U.S.A.*, 30, p. 244
13. Macdonald, J.R., Powell, D.R., (1971) *J. Res. Natl. Bur. Stand., Sect. A*, 75, p. 441
14. Duclos, S.J., (1991) *Nature* (London), 351, p. 380
15. Leger, J.M., (2002) *Solid State Commun.*, 121, p. 241
16. Kawasaki, S., (2003) *Solid State Commun.*, 125, p. 637
17. Serebryanaya, N.N., (2001) *Solid State Commun.*, 118, p. 183
18. note Ozaki, T., Iwasa, Y., Mitani, T., (1998) *Chem. Phys. Lett.*, 285, p. 289
19. Marques, L., (1999) *Science*, 283, p. 1720
20. Chernozatonskii, L.A., Serebryanaya, N.R., Mavrin, B.N., (2000) *Chem. Phys. Lett.*, 316, p. 199
21. Larson, A.C., Von Dreele, R.B., (1998) *General Structural Analysis System*, , Los Alamos National Laboratory