

Polymethylthiophene/Nafion-modified glassy carbon electrode for selective detection of dopamine in the presence of ascorbic acid

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Abstract: The possible use of an electrode modified with electroactive conductive poly(3-methylthiophene) (PMeT)/Nafion as a chemical sensor was investigated for the voltammetric analysis of Dopamine (DA). The electrochemical behavior of dopamine was examined by cyclic voltammetry (CV) and differential pulse voltammetry (DPV) techniques. By using a PMeT-modified glassy carbon (GC/PMeT) electrode, DA and Ascorbic Acid (AA) signals could be separated but the AA at high concentrations still caused significant interference by overlapping the DA peak. In comparison to the GC/PMeT electrode, the glassy carbon (GC/Nafion/PMeT) electrode modified with hybrid film Nafion/PMeT was found to permit a superior separation by shifting the oxidation of AA peak toward the less positive potential. The DPV curves for a mixture of DA and AA at an GC/Nafion/PMeT electrode in a 0.1 M H₂SO₄ solution showed peaks of DA and AA, at 0.45 and 0.21 V, respectively, indicating that the difference in the oxidation potential was 240 mV. In the 0.1 M H₂SO₄ solution, the oxidation peak current on the differential pulse voltammograms for the GC/PMeT electrode increased linearly with the concentration of DA in the range 1×10^{-6} to 1×10^{-3} M, and the oxidation peak current on the differential pulse voltammograms for the GC/Nafion/PMeT electrode in the range 5×10^{-7} to 2×10^{-4} M. The DA detection sensitivity of the GC/Nafion/PMeT electrode ($26.7 \text{ } \mu\text{A } \mu\text{M}^{-1} \text{ cm}^{-2}$) was 22 times higher than that of the GC/PMeT electrode ($1.21 \text{ } \mu\text{A } \mu\text{M}^{-1} \text{ cm}^{-2}$). © 2009 Springer Science+Business Media B.V.

Author Keywords: Ascorbic acid; Dopamine; Electrochemical; Nafion; Polymethylthiophene

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References:

1. Mo, J.-W., Ogorevc, B., Simultaneous measurement of dopamine and ascorbate at their physiological levels using voltammetric microprobe based on overoxidized poly(1,2- phenylenediamine)-coated carbon fiber (2001) *Analytical Chemistry*, 73 (6), pp. 1196-1202. , DOI 10.1021/ac0010882
2. Wightman, R.M., May, L.J., Michael, A.C., (1988) *Anal Chem*, 60, pp. 769A. , 10.1021/ac00164a001 1:CAS:528:DyaL1cXksVymsLg%3D
3. Wang, H.Y., Sun, Y., Tang, B., Study on fluorescence property of dopamine and determination of dopamine by fluorimetry (2002) *Talanta*, 57 (5), pp. 899-907. , DOI 10.1016/S0039-9140(02)00123-6, PII S0039914002001236
4. Hard, D.L., Bhatnagar, R.K., Molina, J.R., Anderson, L.L., Secretion of dopamine and norepinephrine in hypophyseal portal blood and prolactin in peripheral blood of Holstein cattle (2001) *Domestic Animal Endocrinology*, 20 (2), pp. 89-100. , DOI 10.1016/S0739-7240(01)00085-6, PII S0739724001000856
5. Cannazza, G., Di Stefano, A., Mosciatti, B., Braghiroli, D., Baraldi, M., Pinnen, F., Sozio, P., Parenti, C., Detection of levodopa, dopamine and its metabolites in rat striatum dialysates following peripheral administration of L-DOPA prodrugs by mean of HPLC-EC (2005) *Journal of Pharmaceutical and Biomedical Analysis*, 36 (5), pp. 1079-1084. , DOI 10.1016/j.jpba.2004.09.029, PII S0731708504004431
6. Atta, N.F., Galal, A., Karag??zler, A.E., Russell, G.C., Zimmer, H., Mark, H.B.J., (1991) *Biosens Bioelectron*, 6, p. 333. , 10.1016/0956-5663(91)85020-W 1:CAS:528:DyaK38XivFKj
7. Zucolotto, V., Ferreira, M., Cordeiro, M.R., Constantino, C.J.L., Moreira, W.C., Oliveira Jr., O.N., Nanoscale processing of polyaniline and phthalocyanines for sensing applications (2006) *Sensors and Actuators, B: Chemical*, 113 (2), pp. 809-815. , DOI 10.1016/j.snb.2005.03.114, PII S0925400505003497

8. Mathiyarasu, J., Senthilkumar, S., Phani, K.L.N., Yegnaraman, V., Selective detection of dopamine using a functionalised polyaniline composite electrode (2005) *Journal of Applied Electrochemistry*, 35 (5), pp. 513-519. , DOI 10.1007/s10800-005-0914-6
9. Chen, J., Cha, C.S., (1999) *J Electroanal Chem*, 463, p. 93. , 10.1016/S0022-0728(98)00451-3
1:CAS:528:DyaK1MXhslOitro%3D
10. Zhang, Y., Cai, Y., Su, S., Determination of dopamine in the presence of ascorbic acid by poly(styrene sulfonic acid) sodium salt/single-wall carbon nanotube film modified glassy carbon electrode (2006) *Analytical Biochemistry*, 350 (2), pp. 285-291. , DOI 10.1016/j.ab.2006.01.002, PII S0003269706000054
11. Zhao, H., Zhang, Y., Yuan, Z., Electrochemical determination of dopamine using a poly(2-picolinic acid) modified glassy carbon electrode (2001) *Analyst*, 126 (3), pp. 358-360. , DOI 10.1039/b008283j
12. Erdogdu, G., Karagozler, A.E., Investigation and comparison of the electrochemical behavior of some organic and biological molecules at various conducting polymer electrodes (1997) *Talanta*, 44 (11), pp. 2011-2018. , DOI 10.1016/S0039-9140(96)02196-0, PII S0039914096021960
13. Zheng, J., Zhou, X., (2006) *Bioelectrochemistry*, 71, p. 106
14. Bouchta, D., Izaoumen, N., Zejli, H., El Kaoutit, M., Temsamani, K.R., A novel electrochemical synthesis of poly-3-methylthiophene-??- cyclodextrin film: Application for the analysis of chlorpromazine and some neurotransmitters (2005) *Biosensors and Bioelectronics*, 20 (11), pp. 2228-2235. , DOI 10.1016/j.bios.2004.12.004
15. Wang, H.-S., Li, T.-H., Jia, W.-L., Xu, H.-Y., Highly selective and sensitive determination of dopamine using a Nafion/carbon nanotubes coated poly(3-methylthiophene) modified electrode (2006) *Biosensors and Bioelectronics*, 22 (5), pp. 664-669. , DOI 10.1016/j.bios.2006.02.007, PII S0956566306000583
16. Raoof, J.-B., Ojani, R., Rashid-Nadimi, S., Voltammetric determination of ascorbic acid and dopamine in the same sample at the surface of a carbon paste electrode modified with polypyrrole/ ferrocyanide films (2005) *Electrochimica Acta*, 50 (24), pp. 4694-4698. , DOI 10.1016/j.electacta.2005.03.002, PII S0013468605001982
17. Hocevar, S.B., Wang, J., Deo, R.P., Musameh, M., Ogorevc, B., Carbon nanotube modified microelectrode for enhanced voltammetric detection of dopamine in the presence of ascorbate (2005) *Electroanalysis*, 17 (5-6), pp. 417-422. , DOI 10.1002/elan.200403175
18. Wu, K., Hu, S., Electrochemical study and selective determination of dopamine at a multi-wall carbon nanotube-nafion film coated glassy carbon electrode (2004) *Microchimica Acta*, 144 (1-3), pp. 131-137. , DOI 10.1007/s00604-003-0103-4
19. Zhang, M., Gong, K., Zhang, H., Mao, L., Layer-by-layer assembled carbon nanotubes for selective determination of dopamine in the presence of ascorbic acid (2005) *Biosensors and Bioelectronics*, 20 (7), pp. 1270-1276. , DOI 10.1016/j.bios.2004.04.018, PII S0956566304002052
20. Shimanouchi, T., Morita, S., Jung, H.S., Sakurai, Y., Suzuki, Y., Kuboi, R., (2004) *Sensor Mater*, 16, p. 255. , 1:CAS:528:DC%2BD2cXhtFWqur%2FL
21. Jung, H.S., Shimanouchi, T., Morita, S., Kuboi, R., (2003) *Electroanalysis*, 15, p. 1453. , 10.1002/elan.200302660
1:CAS:528:DC%2BD3sXosVKjsL4%3D
22. Sakurai, Y., Jung, H.-S., Shimanouchi, T., Inoguchi, T., Morita, S., Kuboi, R., Natsukawa, K., Novel array-type gas sensors using conducting polymers, and their performance for gas identification (2002) *Sensors and Actuators, B: Chemical*, 83 (1-3), pp. 270-275. , DOI 10.1016/S0925-4005(01)01069-3, PII S0925400501010693
23. Tuckerman, M.M., Mayer, J.R., Nachod, F.C., (1959) *J Am Chem Soc*, 81, p. 92. , 10.1021/ja01510a020
1:CAS:528:DyaG1MXmtlektA%3D%3D
24. Heitner-Wirguin, C., Recent advances in perfluorinated ionomer membranes: Structure, properties and applications (1996)

Journal of Membrane Science, 120 (1), pp. 1-33. , DOI 10.1016/0376-7388(96)00155-X

25. Rocha, L.S., Carapuca, H.M., Ion-exchange voltammetry of dopamine at Nafion-coated glassy carbon electrodes: Quantitative features of ion-exchange partition and reassessment on the oxidation mechanism of dopamine in the presence of excess ascorbic acid (2006) *Bioelectrochemistry*, 69 (2), pp. 258-266. , DOI 10.1016/j.bioelechem.2006.03.040, PII S1567539406000648
26. Selvaraju, T., Ramaraj, R., (2005) *J Electroanal Chem*, 585, p. 290. , 10.1016/j.jelechem.2005.09.005
1:CAS:528:DC%2BD2MXht1SjsrjJ
27. Jaiswal, P.V., Ijeri, V.S., Srivastava, A.K., Effect of surfactants on the dissociation constants of ascorbic and maleic acids (2005) *Colloids and Surfaces B: Biointerfaces*, 46 (1), pp. 45-51. , DOI 10.1016/j.colsurfb.2005.09.001, PII S0927776505002730